

2.  **$\beta$  -particles:** The fast-moving stream of electrons that are produced during the conversion of a neutron into a proton is called  $\beta$  rays.

The  $\beta$  -particle is not present initially in the nucleus but it is produced due to the conversion of a neutron into a proton.

➤ The charge and mass of a **beta particle** is same as that of an electron.

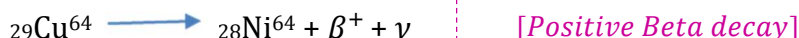
*A beta particle is an electron of nuclear origin.*

a. **Negative  $\beta$  -particle:** A fast moving electron emitted from the nucleus during radioactivity. The *Negative  $\beta$  -particle* is not present initially in the nucleus but it is produced due to the conversion of a neutron into a proton.



b. **Positive  $\beta$  -particle:** A fast moving positron (antiparticle of electron) emitted from the nucleus during radioactivity.

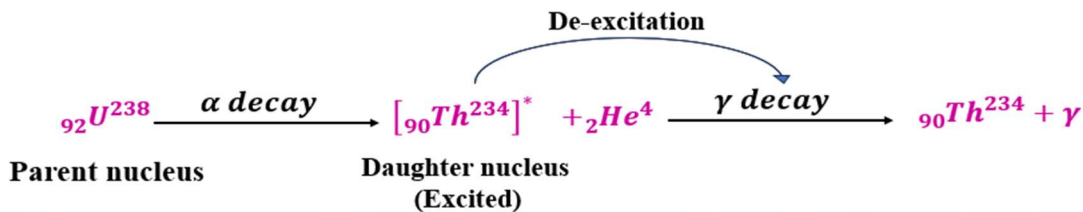
The *Positive  $\beta$  -particle* is not present initially in the nucleus but it is produced due to the conversion of a proton into a neutron.



3.  **$\gamma$  -rays:** The electromagnetic waves having very short wavelength of the order of  $0.005\text{\AA}$  to  $0.5\text{\AA}$  are called  $\gamma$  - rays.

After  $\alpha$  or  $\beta$  decay, a new nucleus is formed (called as daughter nucleus). This daughter nucleus usually remains in excited state. During the de-excitation of the nucleus, it loses energy in the form of  **$\gamma$  radiation**.

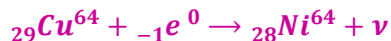
*Hence, gamma decay always occurs after alpha or beta decay.*



- Alpha and beta decays of a radioactive nucleus usually leave the daughter nucleus in an **excited state**.
- If the excitation energy available with the daughter nucleus is not sufficient for further particle emission, it loses its energy by emitting electromagnetic radiations, also known as  **$\gamma$  -rays**.
- Atomic mass and charge of the daughter nucleus **remains the same** before and after the emission of  **$\gamma$  - rays**. Thus  **$\gamma$  - rays** are electromagnetic waves having no mass and no charge.

#### Electron capture:

Here the parent nucleus captures one of the orbital electrons with the emission of a neutrino ( $\nu$ ). Mass number of the daughter nucleus remains the same and atomic number decreases by one unit. For example,



- ✚ Write the properties of alpha, beta and gamma particles.
- ✚ Alpha particles have the maximum ionizing power but minimum penetrating power. Explain.